New for 2022 – 2023
Test plan percentages have been added for each event. The ILC Registration process has been clarified.
HOSA is pleased to announce that all eleven (11) the National Geographic Learning Academic Testing Center events will be available for chartered associations to use at State Leadership Conferences (in addition to being offered at the International Leadership Conference) at the discretion of the state. Contact your State Advisor to determine if the tests will be offered at your SLC.

Event Summary
The goal of the National Geographic Learning Academic Testing Center (NGL ATC) Cengage sponsored events is to provide as many International Leadership Conference HOSA delegates as space permits with the opportunity to demonstrate their basic knowledge in preparation to become future health professionals.

The series of events in the NGL ATC are written tests based on items from the identified text specific to each event. Competitors will recognize, identify, define, interpret, and apply knowledge in a 50-item multiple choice test with a tie-breaker question. The written test will measure knowledge and understanding at the recall, application, and analysis levels. Higher-order thinking skills will be incorporated.

Sponsorship. These competitive events are sponsored by National Geographic Learning / Cengage

Dress Code
Competitors must be in official HOSA uniform or in proper business attire. Bonus points will be awarded for proper dress.

Competitor Must Provide:
- Photo ID
- Two #2 lead pencils (not mechanical) with eraser
Rules and Procedures

1. Competitors in this event must be active members of HOSA-Future Health Professionals and in good standing.

2. **Eligible Divisions:** Secondary, Postsecondary/Collegiate, or Alumni division members may compete in this event.

3. Competitors must be familiar with and adhere to the “General Rules and Regulations of the HOSA Competitive Events Program (GRR).”
   - Per the GRRs #11 and Appendix H, HOSA members may request accommodation in any competitive event. To learn the definition of an accommodation, please read Appendix H. To request accommodation for the International Leadership Conference, submit the request form here by May 15 at midnight EST.
   - To request accommodation for any regional/state level conferences, please work with your local and state advisor directly. Accommodations must first be done at state in order to be considered for ILC.

4. A tie-breaker question(s) will be administered with the original test. In case of a tie, the tie-breaker question(s) will be graded to break the tie.

5. All competitors shall report to the site of the event at the time designated for the test in ILC publications. Competitors will bring a photo ID as well as two #2 lead pencils (not mechanical) for test-taking.

6. **Test Instructions:** The competitors will be given instructions and will be notified to start the test. There will be a maximum of 60 minutes to complete the test.

7. **TIME REMAINING ANNOUNCEMENTS:** There will be NO verbal announcements for time remaining during ILC testing. All ILC testing will be completed in the Testing Center and competitors are responsible for monitoring their own time.

Testing Areas and Resources

8. Eleven (11) of the NGL Academic Testing Center events are sponsored by National Geographic Learning / Cengage. A summary page of resources for these 11 NGL ATC events is available, as well as a listing of all resources provided from National Geographic Learning / Cengage for all HOSA events.

9. The eleven (11) approved tests for ILC and their associated resources are as follows:

<table>
<thead>
<tr>
<th>Event</th>
<th>Resource Title (National Geographic Learning/Cengage)</th>
<th>ISBN</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Allied Health Statistics</td>
<td>Basic Allied Health Statistics and Analysis, 5th Edition</td>
<td>9781337796965</td>
</tr>
<tr>
<td>2. Anatomy and Physiology</td>
<td>Body Structures and Functions, 14th edition,</td>
<td>9780357457542</td>
</tr>
<tr>
<td>3. Biochemistry</td>
<td>Introduction to General, Organic and Biochemistry, 12th edition</td>
<td>9781337571357</td>
</tr>
<tr>
<td>4. Biology</td>
<td>Biology: The Unity and Diversity of Life, 15th AP Edition</td>
<td>9781337408592</td>
</tr>
<tr>
<td>5. Career Development</td>
<td>DHO Health Science Updates 9th Edition</td>
<td>9780357419991</td>
</tr>
</tbody>
</table>
10. Test Plans:

**Allied Health Statistics:**
- Identify the importance of health statistics 12%
- Describe health data across the continuum 8%
- Describe hospital census information 8%
- Identify hospital occupancy information 8%
- Understand statistics related to hospital length of stay 8%
- Understand hospital obstetric and neonatal statistics 8%
- Identify clinical and non-clinical statistics 8%
- Describe mortality and autopsy rates 8%
- Understand community health statistics 5%
- Understand the basics of statistics 5%
- Describe organizing data for analysis 8%
- Identify how to display data for analysis 6%
- Understand the fundamentals of research 8%

**Anatomy & Physiology:**
- Compare and contrast anatomy and physiology 2%
- Describe the physiology of the human body 37%
- Identify the anatomy of the human body 37%
- Identify basic medical terminology word elements and combining forms 8%
- Identify common diseases and disorder of the human body 12%
- Understand basic medical math problems 4%

**Biology:**
- Describe the study of life and organization 6%
- Identify the building blocks of matter 4%
- Describe the structures and functions of plant and animal cells 18%
- Describe the mechanisms of genetics and heredity 14%
- Identify how hormones work in plants and animals 10%
- Describe evolutionary adaptations in plants and animals 8%
- Identify structures and functions of plant and animal systems 14%
- Identify plant and animal reproduction processes 16%
- Describe how climate, community, and populations affect plants and animals 10%

**Biochemistry:**
- Understand the biochemical reactions that occur in water 10%
- Describe the organization of cells 8%
- Describe amino acids and peptides 6%
- Identify proteins, protein purification, and enzymes 14%
- Describe biological membranes 6%
• Understand how structures of nucleic acids conveys information 3%
• Describe the biosynthesis and replication of nucleic acids 8%
• Understand the biosynthesis of RNA 4%
• Describe the translation of genetic messages 6%
• Understand viruses, cancer, and immunology 6%
• Describe the importance of energy and electron transfer in metabolism 10%
• Identify storage mechanisms in carbohydrate metabolism 16%
• Describe glycolysis 3%

Career Development:
• Recognize how to begin the process of career development 10%
• Identify the required skills needed for career development 12%
• Describe individual preferences for career development 8%
• Identify values needed for career development 8%
• Understand the exploration stage of career development 10%
• Describe the relationships required for career development 10%
• Identify the stages of the decision making 14%
• Determine the tools required for career development 6%
• Understand the launch stage of career development 16%
• Recognize the skills required for career management 6%

General Chemistry:
• Describe chemistry concepts, measurements, and calculations 13%
• Identify elements, atoms, and ions 6%
• Describe chemical reactions, energy, and solutions 31%
• Compare solids, liquids, and gases 12%
• Identify chemical composition and quantities 12%
• Understand chemical bonding 6%
• Identify modern atomic theory 4%
• Compare and contrast acids and bases 6%
• Understand principles of chemical nomenclature 6%
• Identify the properties of matter 4%

Human Heredity:
• Identify common genetic diseases and disorders 26%
• Understand genetics, genes, genomes, and gene therapies 15%
• Understand chromosomes, inheritance, and the transmission of traits 15%
• Describe how immunity is related to genetics 2%
• Describe DNA, the structure of DNA, and how it relates to genetics 8%
• Understand the process and stages of meiosis 6%
• Describe mutations and deletions 10%
• Describe human reproduction 4%
• Identify genomic technology 2%
• Understand RNA and how it relates to genetics 2%
• Understand how amino acids influence genetics 2%
• Describe the process of cloning 4%
• Identify how viruses can influence genetics 2%

Math for Health Professionals:
• Understand how to create the proper solution 20%
• Understand basic mathematical concepts and conversions 21%
• Understand how to determine proper drug dosages 30%
• Understand how to measure input and output 3%
• Identify how to convert roman numerals 3%
• Identify how to determine drops per minute and flow rate 11%
• Identify how to convert from the metric system to the English system and vice versa 10%
• Understand basic medical terminology 2%

Microbiology:
• Identify the different types of sanitation, sterilization, and disinfection 19%
• Describe ways that pathogens can be transmitted 8%
• Identify diseases and disorders 21%
• Understand the Chain of Infection 4%
• Identify healthcare associated infections 8%
• Understand the history of microbiology 2%
• Describe the types of immunity 12%
• Identify tests that can be used to determine infection 4%
• Understand asexual reproduction 2%
• Identify the characteristics of mold 2%
• Understand classifications of organisms 4%
• Identify characteristics of prokaryotic cells 2%
• Describe the different types of stains 4%
• Understand the pH scale 2%
• Describe the characteristics of bacteria 6%

Organic Chemistry:
• Compare and contrast chemical reactions 16%
• Describe nuclear magnetic, infrared, and mass spectroscopy 11%
• Identify the IUPAC names of common compounds 11%
• Describe amino, nucleic, and carboxylic acids 14%
• Identify chemical compounds and products 16%
• Identify chemical bonds and molecular shapes 8%
• Compare and contrast polymers 2%
• Describe substitutions 6%
• Identify acids and bases 4%
• Describe alkanes, cycloalkanes, and alcohols 4%
• Identify stereoisomerism and chirality 4%
• Describe ethers, epoxides, and sulfides 4%

Physics:
• Understand radio waves, frequency, and wavelength 5%
• Identify resistance, watts, and amps 4%
• Describe the speed of decay 2%
• Understand the principles of buoyancy 2%
• Describe the relationship between temperature and humidity 2%
• Understand the concept of entropy 2%
• Identify different electrical charges, electrical circuits, and electrical fields 20%
• Understand the history of radiation 2%
• Compare and contrast different light rays and their characteristics 12%
• Understand velocity, acceleration, and force 20%
• Describe images and focal points 5%
• Understand the concept of magnetism 10%
• Understand basic mathematical concepts related to physics 10%
• Describe thermal dynamics 4%

11. Chapter Advisors – contact your Cengage Sales Consultant for a review copy of the reference title of your choice, for events 1-11 above. Most resources also offer online interactive study tools, virtual labs, and auto-graded practice. Find your Sales Consultant at NGL.Cengage.com/repfinder
Registration Priority and Process

12. **ILC registration** includes the opportunity to take one NGL Academic Testing Center test. Delegates who wish to take additional NGL Academic Testing Center tests may do so by paying $20 per additional test. There is no maximum to the number of NGL ATC tests that can be taken at ILC, as long as the competitor can logistically sit to test in all events for which they are registered.

13. Delegates will pre-register for the NGL Academic Testing Center at ILC in the HOSA Conference Management System. No walk-up NGL ATC registration is allowed at ILC, only registered delegates by the published ILC deadline will be allowed to test.

14. Competitors should refer to the General Rules and Regulations #6 for information on how many competitive events they can register for at ILC.

15. At State Chartered Association Conferences, these eleven (11) tests **MAY** be available as an additional event opportunity. Please check with your state HOSA leadership for details.

Recognition and Awards

16. A Certificate of Participation is given to every delegate who takes a test in the NGL Academic Testing Center.

17. The Top Ten HOSA members in each event will be recognized on stage at the Grand Awards Ceremony, with 1st, 2nd and 3rd place receiving special recognition.

18. HOSA Advisors will receive a Certificate of Excellence for any HOSA members from their chapter ranking in the Top Ten.

Sample Test Questions

The remaining pages of these guidelines include sample test questions from each of the eleven event tests.

Allied Health Statistics

1. Which of the following facilities delivers the highest level of nursing care?
   a. Intermediate Care
   b. **Residential Care**
   c. Skilled Nursing Care
   d. Rehabilitation Care

2. Which type of validity refers to how well the study adheres to accepted and established standards?
   a. Face validity
   b. Content validity
   c. **Construct validity**
   d. Criterion validity

3. Where do low numbers appear in a frequency polygon?
   a. Bottom and to the right
   b. Bottom and to the left
   c. Top and to the right
   d. **Top and to the left**
Anatomy and Physiology
1. Health care workers use a spirometer to measure the_________.
   a. level of carbon dioxide in the blood
   b. lungs' capacity for air
   c. amount of pressure in the lungs
   d. level of pleural fluid

2. What causes plasma like fluid to flow from the blood in the glomerulus into Bowman’s capsule?
   a. Increase in blood pressure in the capillaries
   b. Hormonal secretions
   c. Stimulation from the nerves
   d. Level of salt in the blood

3. Which of the following is the definition of a fomite?
   a. Bacterial infection transmitted through contaminated water
   b. Small worm that may be present in meat and which infects the intestinal tract.
   c. Person who experiences no symptoms but can transmit an infection.
   d. Nonliving object that is contaminated with an infectious agent

Biochemistry
1. Which of the following compounds would have the highest boiling point?
   a. CH₃CH₂CH₂CH₃
   b. CH₃NH₂
   c. CH₃OH
   d. CH₂F₂

2. Which subatomic particle is found in all isotopes of hydrogen?
   a. Proton
   b. Neutron
   c. Electron
   d. Positron

3. CH₃C ≡ CCH₂CH₂Cl is named:
   a. 1-chloro-3-pentyne
   b. 5-chloro-2-pentene
   c. 1-acetylenyl-3-chloropropane
   d. 5-chloro-2-pentyne

Biology
1. In the electromagnetic spectrum,_________.
   a. infrared energy has the shortest wavelength
   b. infrared radiation has more energy than red radiation
   c. visible light provides the energy for photosynthesis
   d. near-infrared radiation provides the energy for photosynthesis

2. In order for DNA molecules to undergo recombination,_________.
   a. they must be from the same species
   b. their strands must separate as in replication
   c. they must be cut and spliced at specific nucleotide sequences
   d. one of the two DNA strands must be degraded
3. In sweet peas, dominant alleles $C$ and $P$ are necessary for colored flowers. In the absence of either ($__ pp$ or $cc __$) or both ($cc pp$), the flowers are white. What are the phenotypic ratios for the cross of $Cc pp \times cc Pp$?
   a. 1/2 color, 3/8 white
   b. 9/16 color, 7/16 white
   c. 3/4 color, 1/4 white
   d. 1/4 color, 3/4 white

Career Development
1. If a physician fails to use a degree of skill and learning commonly expected and the person receiving care is injured, the physician can be sued for ______.
   a. Negligence
   b. Defamation
   c. Malpractice
   d. Assault and battery

2. Patients confined to bed should have their position changed at least every ______.
   a. 30 minutes
   b. Hour
   c. 2 hours
   d. 3 hours

3. $5/8 + 3/12 = _____.
   a. 11/12
   b. 13/16
   c. 8/24
   d. 7/8

General Chemistry
1. Select the correct molecular structure for CO$_2$:
   a. linear
   b. trigonal planar
   c. tetrahedral
   d. Bent

2. Consider the molecular orbital description of the NO$^-$ anion. Which of the following statements is false?
   a. NO$^-$ is paramagnetic.
   b. NO$^-$ is isoelectronic with CO.
   c. The bond energy in NO$^+$ is greater than the bond energy in NO$^-$.  
   d. The bond order in NO$^-$ is 2.

3. For which of the following compound(s) are cis and trans isomers possible?
   a. 2,3-dimethyl-2-butene
   b. 3-methyl-2-pentene
   c. 4,4-dimethylcyclohexanol
   d. orthochlorotoluene

Human Heredity
1. The letters G, Q, R, and C, used to describe the appearance of chromosomes, refer to the ___.
   a. position of the bands
   b. staining procedure used to reveal the bands
   c. number of arms per chromosome
   d. number of centromeres per chromosome
2. Which of the following sequences indicates the promoter region of a gene?
   a. CAAT
   b. UAAG
   c. CTTT
   d. ACAT

3. The ability to taste PTC and other bitter chemicals is controlled by__________.
   a. hormone levels that change throughout life
   b. **proteins on the surface of receptor cells**
   c. the amount of PTC exposure as a child
   d. the amount of capsaicin present in taste buds

**Math for Health Professionals**
1. Goniometers measure
   a. range of motion
   b. weight
   c. surface area
   d. Time

2. The hospital uses a 15-drop-per-mL drip set. How will you adjust the IV to infuse 250 mL over 3 hours?
   a. constant infusion
   b. every minute
   c. every 30 seconds
   d. **every 3 seconds**

3. Calculate the BMI for Weight: 165 pounds; Height: 73 inches.
   a. 30
   b. 25
   c. 18
   d. **22**

**Microbiology**
1. By what mechanism does a virus cause disease?
   a. by infecting the nervous system in humans
   b. **by shutting down or destroying a cell**
   c. by living on or in another organism
   d. by infecting red blood cells

2. Toxic proteins that can be secreted outside of the cell are called?
   a. bactericides
   b. endotoxins
   c. **exotoxins**
   d. aspergillus

3. Which main class of disease-causing parasites contain tapeworms?
   b. **Helminths**
   c. protozoa
d. ectoparasites
e. Ergosterol

**Organic Chemistry**
1. Name the following: CH₃–CH₂–CH₃
   a. ethane
   b. **propane**
   c. butane
   d. Pentane
2. \( \text{CH}_3\text{C} \equiv \text{CCH}_2\text{CH}_3\text{Cl} \) is named:
   a. 1-chloro-3-pentyne
   b. 5-chloro-2-pentene
   c. 1-acetylenyl-3-chloropropane
   d. 5-chloro-2-pentyne

3. What is the correct (IUPAC) name of the following molecule: 2-methyl-4-t-butylpentane
   a. 2-t-buty1-4-methylpentane
   b. \( 2,2,3,5 \)-tetramethylhexane
   c. 2,4,5,5-tetramethylhexane
   d. 1-sec-butyl-1,2,2-trimethylpentane

Physics College
1. How many moles of air must escape from a 15.0-m \( \times \) 9.0-m \( \times \) 6.0-m room when the temperature is raised from 10.0°C to 20.0°C? Assume the pressure remains unchanged at one atmosphere while the room is heated. \((R = 8.31 \text{ J/mol-K})\)
   a. 4.9E+3 moles
   b. \( 1.2E+3 \) moles
   c. 2.2E+2 moles
   d. 7.9E+2 moles

2. A loop of area 0.384 m\(^2\) is in a uniform 0.0565-T magnetic field. If the flux through the loop is 6.10 \( \times \) 10\(^{-3}\) T\(\cdot\)m\(^2\), what angle does the normal to the plane of the loop make with the direction of the magnetic field?
   a. 73.7°
   b. 89.3°
   c. 16.3°
   d. 76.0°

3. The escape speed from the surface of the Earth is 11.2 km/s. Estimate the escape speed for a spacecraft from the surface of the Moon. The Moon has a mass 1/81 that of Earth and a radius 0.25 that of Earth.
   a. 2.5 km/s
   b. 4.0 km/s
   c. 5.6 km/s
   d. 8.7 km/s